

BENZOFURANS FROM *CLIBADIUM ARMANII*

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(Revised received 29 February 1980)

Key Word Index—*Clibadium armanii*; *C. microcephalum*; Heliantheae; Compositae; new benzofurans; acetophenone derivatives.

The roots of *C. armanii* Sch. Bip, ex Baker collected in north-eastern Brazil afforded, in addition to squalene, heptadeca-7,9,15-triene-11,13-diyne (**1**) [1], tetradeca-4,6,12-triene-8,10-diyne (**2**) [1], lachnophyllumester (**3**) [1], 6-hydroxytremetone (**4**) [2] and 6-hydroxytoxol (**5**), isolated so far only from *Fitchia speciosa* [3], and two further acetophenone derivatives. In both of the latter compounds the isopropylidene group was replaced by an isopropyl group. The more polar compound **7** was obviously a 10,11-dihydro derivative of **5**, as could be seen by the similar ¹H NMR spectral data of both compounds (Table 1). As expected the 2-H signal was shifted upfield. The second compound was the diketone **6**. Again the ¹H NMR spectral data (Table 1) clearly showed that the second keto group could only be placed at C-3. Consequently the proton at C-4 was shifted downfield in comparison with the chemical shifts in the spectra of **4**, **5** and **7**. So far no compounds of this type have been previously isolated. The aerial parts of the plant contain **1**,

2, **4** and squalene as well as germacrene D (**8**) and bicyclogermacrene (**9**).

The roots of *C. microcephalum* Blake, however, afforded trideca-1,11-diene-3,5,7,9-tetrayne (**10**) [1] and linolenic alcohol (**11**), while the aerial parts gave phytol and α -copaene (**12**) in addition to **10** and **11**. The chemotaxonomic situation of this genus is still complicated. Some species show chemistry similar to *Ichthyothere* [4, 5], which is placed in the same subtribe [6], while others have constituents of different types, which are not very characteristic [5].

EXPERIMENTAL

The air-dried plant material, collected in north-eastern Brazil was chopped and extracted with Et₂O–petrol (1:2). The resulting extracts were first separated by CC (Si gel) and further by TLC (Si gel, GF 254). Known compounds were identified by comparison of the IR and ¹H NMR spectra with those of authentic material.

Clibadium armanii (voucher RMK 8000). The roots (150 g) afforded 5 mg **1**, 1 mg **2**, 3 mg **3**, 2 mg squalene, 20 mg **4**, 4 mg **5**, 6 mg **6** (Et₂O–petrol, 1:3) and 1 mg **7** (Et₂O–petrol, 1:1), while the aerial parts (250 g) gave 25 mg **1**, 4 mg **2**, 4 mg **4**, 25 mg **8**, 13 mg **9** and 13 mg squalene.

C. microcephalum (voucher RMK 7972). The roots (100 g) afforded 6 mg **10** and 150 mg **11**, while the aerial parts (1250 g) yielded 70 mg **10**, 200 mg **11**, 30 mg **12** and 50 mg phytol.

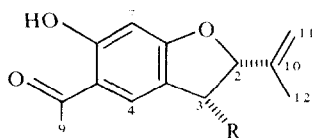
10,11-Dihydro-6-hydroxy-3-oxotremetone (6). Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 3400–2700, 1645 (hydrogen-bonded PhCO); MS m/e (rel. int.): 234.089 (M⁺, 10%) (C₁₃H₁₄O₄); 219 (10, M – 'Me); 192 (100, M – C₃H₆, McLafferty); 177 (25, 192 – 'Me).

10,11-Dihydro-6-hydroxy-toxol (7). Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 3610 (OH), 3500–2800, 1640 (hydrogen-bonded PhCO); MS m/e (rel. int.): 236.105 (M⁺, 42%) (C₁₃H₁₆O₄); 221 (17, M – 'Me); 219 (12, M – 'OH); 203 (12, 221 – H₂O); 173 (100, 203 – CH₂O).

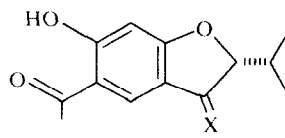
Table 1. ¹H NMR spectral data of compounds **6** and **7** (270 MHz, CDCl₃)

	6	7
2-H	4.53 <i>d</i>	4.06 <i>dd</i>
3-H	—	5.05 <i>d</i> (<i>br</i>)
4-H	8.18 <i>s</i>	7.80 <i>s</i>
7-H	6.60 <i>s</i>	6.39 <i>s</i>
9-H	2.65 <i>s</i>	2.60 <i>s</i>
10-H	2.38 <i>dqq</i>	2.28 <i>dqq</i>
11-H	0.96 <i>d</i>	1.14 <i>d</i>
12-H	1.18 <i>d</i>	1.22 <i>d</i>
OH	13.19 <i>s</i>	12.98 <i>s</i>

$J(\text{Hz})$: **6**: 2, 10 = 4; 10, 11 = 10, 12 = 6.5; **7**: 2, 3 = 5; 2, 10 = 10; 10, 11 = 10, 12 = 6.5.



4 R = H
5 R = OH



6 X = O
7 X = α OH, H

Acknowledgements—We thank Dr. Scott A. Mori and Dr. P. Alvim, Herbario Centro de Pesquisas do Cacau at Itabuna, Bahia, Brazil, for their help during plant collection, and the Deutsche Forschungsgemeinschaft for financial support.

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Phytochemistry, 1980, Vol. 19, pp. 2501–2503. © Pergamon Press Ltd. Printed in England.

0031-9422/80/1101-2501 \$02.00/0

A CHALCONE GLYCOSIDE FROM *ABIES PINDROW*

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(Revised received 29 February 1980)

Key Word Index—*Abies pindrow*: Coniferae: 2',3',4':3,4-pentahydroxy chalcone 4'-(L-arabinofuranosyl- α -1 \rightarrow 4-D-glucopyranoside- β): permethylation: structural determination.

Abstract The EtOH extract of air-dried stems of *Abies pindrow* yielded okanin, okanin 4'-O- β -D-glucopyranoside, butein 4'-O- β -D-glucopyranoside, 8,3',4'-trihydroxyflavanone-7-O- β -D-glucopyranoside and a new chalcone glycoside, 2',3',4':3,4-pentahydroxy-chalcone 4'-(L-arabinofuranosyl- α -1 \rightarrow 4-D-glucopyranoside- β).

INTRODUCTION

Abies pindrow is a tall evergreen tree with a dense conical crown of dark green foliage. Several species of *Abies* are regarded as carminative, expectorant, stomachic, toxic and astringent [1]. Since no work seems to have been previously done on the chemical constituents of *A. pindrow*, we have, therefore, examined the stem of this plant and its constituents are reported in the present paper.

RESULTS AND DISCUSSION

From the EtOH extract of the stem six compounds A to F were isolated. Compound A, a chalcone, has been characterized as okanin by spectral studies, derivative formation, mmp and co-PPC with an authentic sample [2, 3]. Chalcone B could not be studied due to lack of material. Compounds C, D and E were characterized as okanin-4'-O- β -D-glucopyranoside, butein-4'-O- β -D-glucopyranoside and 8,3',4'-trihydroxy-flavanone-7-O- β -D-glucopyranoside, respectively, by comparison of spectral data, mmp and co-PPC with authentic samples [4–7].

Compound F, C₂₆H₃₀O₁₅, gave characteristic tests for a

chalcone-glycoside [8, 9]. On acid hydrolysis it yielded an aglycone, which was identical to compound A in all respects. The sugar portion on co-chromatographic examination revealed the presence of D-glucose and L-arabinose (BuOH–HOAc–H₂O, 4:1:5, *R_f* 0.18 and 0.21 respectively). Quantitative hydrolysis of the glycoside indicated the aglycone content was ca 49%. Quantitative estimation [10] of the sugars present in the hydrolysate revealed that both sugars were present in equimolecular proportions. On partial hydrolysis [11] with 0.02N H₂SO₄ for 7 days at room temperature, it yielded a glycosidic compound which was purified by column chromatography over Si gel using EtOAc as eluant. It was identical with compound C in all respects. The hydrolysate revealed the presence of L-arabinose only, indicating that it was the terminal sugar of compound F. Permethylation [12] of F, followed by hydrolysis showed the presence of 2:3:6-tri-O-methyl-D-glucose and 2:3:5-tri-O-methyl-L-arabinose (BuOH–HOAc–H₂O, 4:1:5, *R_G* 0.83 and 0.95, respectively) [13]. This observation suggested that arabinose was present as the furanoside and that it was attached by C-1 to C-4 of the glucose moiety. The above fact was also confirmed by periodate oxidation [14]. F on